

Orthogonally Protected Thiazole and Isoxazole Diamino Acids: An Efficient Synthetic Route

Jeffrey D. Butler, Keith C. Coffman, Kristin T. Ziebart, Michael D. Toney, and
Mark J. Kurth*^[a]

Heterocyclic and heteroaromatic amino acids (HAAs) are central to the motifs of peptide antibiotics, including microcin B17, nostocyclamide, telomestatin, and thiostrepton.^[1] α -Amino acids undergo cyclization and oxidation to form heteroaromatic rings, notably, thiazoles, oxazoles, indoles, and pyridines, which give rise to well-documented antibiotic activity.^[1] Few of these targets have succumbed to total synthesis due, in large part, to the demand for orthogonally protected HAA building blocks.^[1b] In contrast, commercial orthogonally protected natural amino acids, most commonly lysine and aspartic acid, are routinely used as the branch point in the synthesis of branched or cyclic peptide and oligosaccharide mimetics^[2–6] (Figure 1a). Similarly, these agents see action in the ligation of imaging agents (Fig-

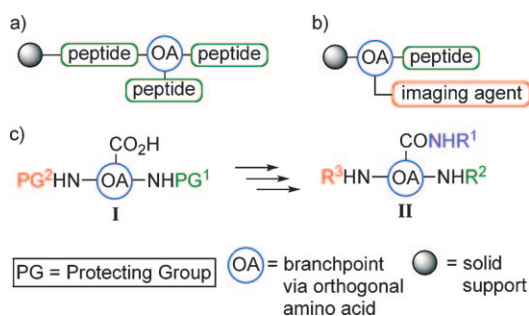


Figure 1. Common motifs and methods employing an orthogonally protected diamino acid, which include a) branched peptides, b) ligated imaging agents, and c) diversity-oriented methods.

ure 1b) and in diversity-oriented syntheses (e.g., **I**→**II**, Figure 1c).^[7,8] However, the stringent orthogonal chemistry requirements, especially in solid-phase synthesis, make optimization at this branch-point region challenging.

Surprisingly, methods to generate new heterocyclic non-natural amino acids with an additional orthogonally protected amino group (e.g., diamino acids), are still rare.^[6,9d] Non-natural conformationally restrictive amino acids have potential in the discovery of new peptidomimetics and in efforts to improve the pharmacological and protease resistant properties of bioactive peptides.^[1b,9] There is demand for practical HAA syntheses that deliver orthogonally protected diamino acids compatible with the traditional solid and solution phase 9-fluorenylmethoxycarbonyl (Fmoc) protection strategy. Thus our focus herein is on the development of short, high yielding syntheses delivering heteroaromatic mono- and diamino acids from readily available starting materials.

Herein, we report an efficient synthesis yielding thiazole- and isoxazole-based HAAs from β -amino acids. This strategy allows for orthogonal carbamate protection that permits independent synthetic manipulation (Figure 1). Further, the viability of the synthesized HAAs as branch-point amino acids is demonstrated in the solid-phase synthesis of an inhibitor of two chorismate utilizing enzymes, anthranilate synthase (AS) and isochorismate synthase (IS). This inhibitor shows two- and threefold better activity than its lysine predecessor in the inhibition of AS and IS, respectively.

A wide variety of β -amino acids are commercially available and considerable synthetic effort has been focused on producing novel optically active β -amino acids.^[10] This availability makes β -amino acids an attractive starting material for this work. As outlined in Figure 2, our synthetic method began by carbamate protection (Teoc, Boc, Cbz, and Alloc) of β -alanine following literature procedures.^[11] These protected acids were subjected to coupling conditions to install the Meldrum acid moiety in 94–98% yield. Intramolecular cyclization of **1a–d**→**2a–d** is accomplished quantitatively in EtOAc at reflux via a presumed ketene intermediate.^[12] In a modification of Suzuki's general method of cyclocondensa-

[a] Dr. J. D. Butler, K. C. Coffman, Dr. K. T. Ziebart, Prof. M. D. Toney, Prof. M. J. Kurth
Department of Chemistry, University of California, Davis
One Shields Avenue, Davis, CA 95616 (USA)
Fax: (+1) 530-752-8995
E-mail: mjkurth@ucdavis.edu
Homepage: <http://chemgroups.ucdavis.edu/~kurth/>

Supporting information for this article is available on the WWW under <http://dx.doi.org/10.1002/chem.201001492>.

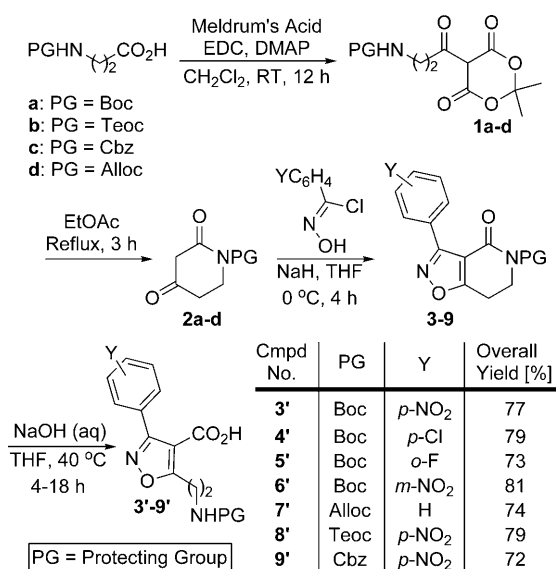


Figure 2. General synthetic method of isoxazole-based amino acids. Boc = *tert*-butoxycarbonyl, Teoc = 2-(trimethylsilyl)ethoxycarbonyl, Cbz = carbobenzyloxy, Alloc = allyloxycarbonyl.

tion,^[13] compounds **2a-d** were treated with strong base (NaH) and α -chlorobenzaldehyde oximes to generate isoxazole adducts **3-9** in 83–90% yield. These mixed imides (**3-9**) were efficiently hydrolyzed to yield amino acids **3'-9'** in 72–81% overall yield without carbamate deprotection.

Our approach to diaminoisoxazole and thiazole HAAs began similarly to the monoamino acids of Figure 2. As delineated in Figure 3, zinc reduction of intermediates **3**, **8**, and **9** (when Y = *p*-NO₂) delivered aniline analogues **10-12**. These anilines were carbamate protected with Alloc and Cbz; subsequent imide hydrolysis delivered a collection of orthogonally protected bis-carbamate isoxazole-based HAAs (**13-17**) in 16–45% overall yield as tabulated in Figure 3.

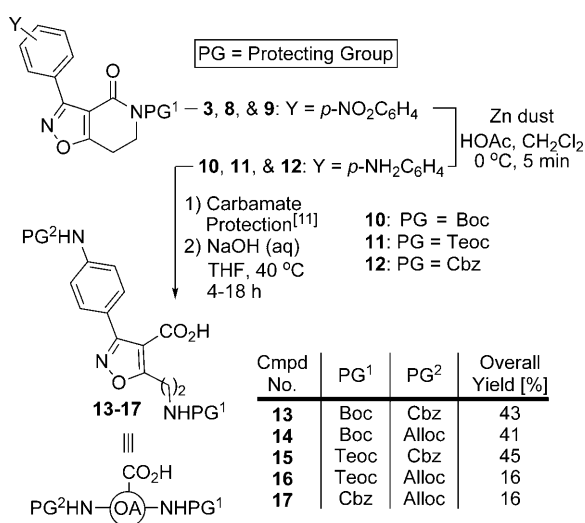


Figure 3. General synthetic method of isoxazole-based orthogonally protected diamino acids.

To deliver orthogonally protected thiazole HAAs, intermediates **1a-c** (Figure 4) were heated in methanol at reflux to quantitatively deliver methyl esters **18a-c**.^[12b] These β -

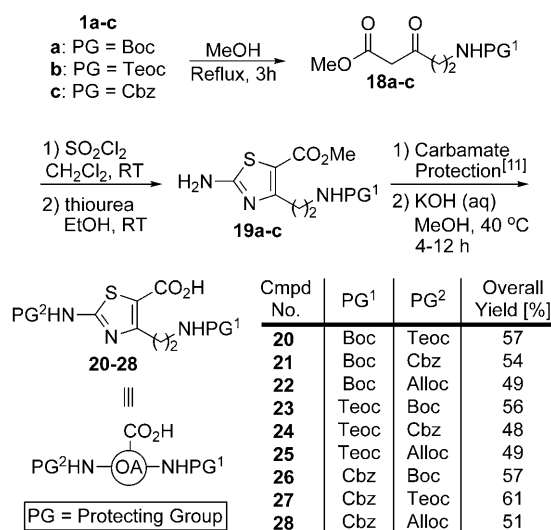


Figure 4. General synthetic method of thiazole-based orthogonally protected diamino acids.

keto esters were α -chlorinated by using sulfonyl chloride and, without purification, condensed with thiourea to give thiazoles (**19a-c**) in two-step yields of 75–79%.^[14] These 2-amino thiazoles were carbamate protected to give orthogonally protected esters that were subsequently saponified to deliver a collection of thiazole-based orthogonally protected diamino acids (**20-28**) as summarized in Figure 4 in an efficient 48–61% overall yield. In addition, these methods have been successful in preparing > 5 g of **25** in comparable overall yield.

To explore how efficiently our diamino HAAs perform in a typical branched synthesis, a solid-phase synthesis of a staged inhibitor analogue (**29**) was undertaken, as depicted in Figure 5. These inhibitors are of particular interest to our group^[8] and are an ideal test case for the employment of these novel orthogonal diamino acids.

To synthesize **29**, Rink amide resin was Fmoc deprotected and *N*-Fmoc-3-chloro-L-phenylalanine was coupled by using standard solid-phase peptide chemistry conditions. After subsequent Fmoc deprotection, the Teoc/Alloc orthogonal HAA **25** was coupled by using DIC and HOBt. Treatment with TBAF in DMF removed the Teoc protecting group^[11] revealing a free amine that was subsequently coupled using dehydrative conditions with 3-hydroxy-4-methyl-2-nitrobenzoic acid. The Alloc protecting group was removed under Pd⁰ conditions.^[11] Finally, DIC- and HOBt-mediated coupling of 3-[(*tert*-butoxycarbonyl)methoxy]benzoic acid completed the independent functionalization of both amines. Resin cleavage and HPLC purification gave **29** in 91% purity and 39% overall yield, which confirms that these or-

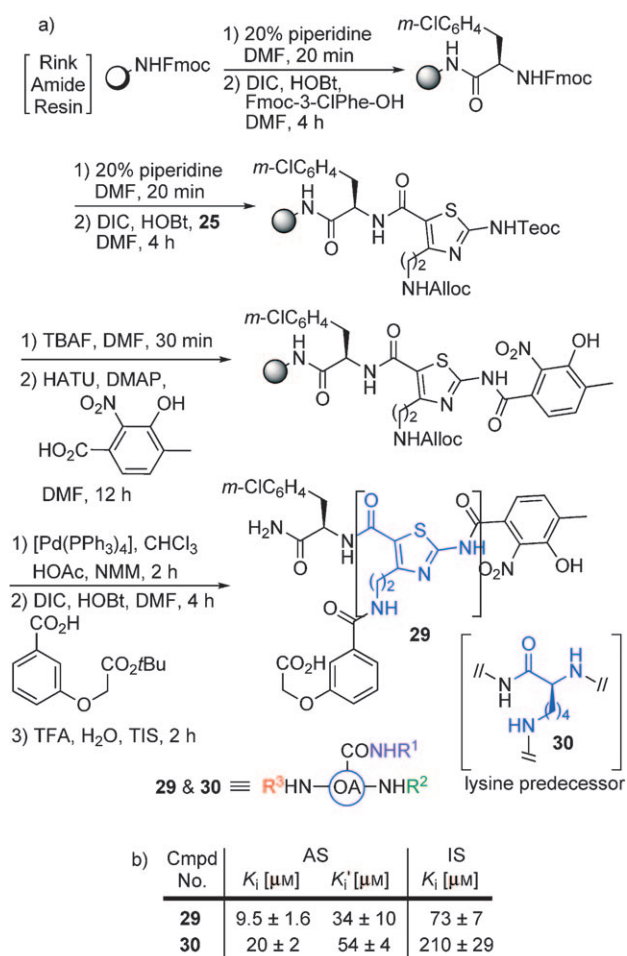


Figure 5. a) Synthetic method of **29** b) IS and AS inhibition data for compounds **29** and **30**. DIC=diisopropylcarbodiimide, HOBT=1-hydroxybenzotriazole, TBAF=tetrabutylammonium fluoride, HATU=O-(7-azabenzotriazol-1-yl)tetramethyluronium hexafluorophosphate, DMAP=4-dimethylaminopyridine, NMM=N-methylmorpholine, TFA=trifluoroacetic acid, TIS=triisopropylsilane.

thogonal diamino HAAs are compatible with standard Fmoc peptide chemistry conditions.

AS and IS are structurally homologous chorismate-utilizing enzymes, and they are excellent antimicrobial drug targets due to their absence in humans and their roles in bacterial and apicomplexan parasite cell survival and/or virulence.^[15] Compound **29** is a structural analogue of a previously discovered inhibitor with a lysine scaffold (**30**).^[8] The inhibition properties of **29** are similar to **30**, but, satisfyingly, the exchange of α-lysine for thiazole HAA gives a significant two- and threefold increase in potency against AS and IS, respectively (Figure 5b). This result represents a step forward in our program to find inhibitors of chorismate-utilizing enzymes. More importantly, this application underscores the value of these HAAs in the development and optimization of tight-binding biological ligands in other discovery efforts.

The goal of developing high-yielding syntheses of heteroaromatic mono- and diamino acids from readily available

starting materials has been realized. Our strategy delivers orthogonally carbamate protected diamino HAAs that are of value for diversity-oriented methods and allow for branched peptide/imaging agent synthesis. Their viability as branch-point amino acids has been successfully demonstrated in the solid-phase synthesis of an inhibitor (**29**) of anthranilate synthase (AS) and isochorismate synthase (IS) with improved potency over its lysine predecessor.

Acknowledgements

We thank the National Science Foundation (CHE-0910870) for generous financial support.

Keywords: amino acids • inhibitors • isoxazoles • protecting groups • thiazoles

- [1] a) H.-D. Arndt, S. Schoof, J.-Y. Lu, *Angew. Chem.* **2009**, *121*, 6900–6904; *Angew. Chem. Int. Ed. Engl.* **2009**, *48*, 6770–6773; b) R. A. Hughes, C. J. Moody, *Angew. Chem.* **2007**, *119*, 8076–8101; *Angew. Chem. Int. Ed.* **2007**, *46*, 7930–7954; c) G. Videnov, D. Kaiser, C. Kemper, G. Jung, *Angew. Chem.* **1996**, *108*, 1604–1607; *Angew. Chem. Int. Ed. Engl.* **1996**, *35*, 1503–1506.
- [2] N. Shepherd, H. Hoang, G. Abbenante, D. Fairlie, *J. Am. Chem. Soc.* **2005**, *127*, 2974–2983.
- [3] S. Di Maro, R. C. Pong, J. T. Hsieh, J. M. Ahn, *J. Med. Chem.* **2008**, *51*, 6639–6641.
- [4] Y. Zheng, K. Balasubramanyam, M. Cebrat, D. Buck, F. Guidez, A. Zelent, R. Alani, P. Cole, *J. Am. Chem. Soc.* **2005**, *127*, 17182–17183.
- [5] S. Pritz, O. Kraetke, A. Klose, J. Klose, S. Rothmund, K. Fechner, M. Bienert, M. Beyermann, *Angew. Chem.* **2008**, *120*, 3698–3701; *Angew. Chem. Int. Ed.* **2008**, *47*, 3642–3645.
- [6] F. Sicherl, V. Wittmann, *Angew. Chem.* **2005**, *117*, 2133–2136; *Angew. Chem. Int. Ed.* **2005**, *44*, 2096–2099.
- [7] a) L. Peng, R. Liu, J. Marik, X. Wang, Y. Takada, K. Lam, *Nat. Chem. Biol.* **2006**, *2*, 381–389; b) R. D. Carpenter, M. Andrei, O. H. Aina, E. Y. Lau, F. C. Lightstone, R. Liu, K. S. Lam, M. J. Kurth, *J. Med. Chem.* **2009**, *52*, 14–19.
- [8] a) S. M. Dixon, K. T. Ziebart, Z. He, M. Jeddelloh, C. L. Yoo, X. Wang, A. Lehman, K. S. Lam, M. D. Toney, M. J. Kurth, *J. Med. Chem.* **2006**, *49*, 7413–7426; b) K. T. Ziebart, S. M. Dixon, B. Avila, M. H. El-Badri, K. G. Guggenheim, M. J. Kurth, M. D. Toney, *J. Med. Chem.* **2010**, *53*, 3718–3729.
- [9] a) A. Giannis, T. Kolter, *Angew. Chem.* **1993**, *105*, 1303–1326; *Angew. Chem. Int. Ed. Engl.* **1993**, *32*, 1244–1267; b) J. Gante, *Angew. Chem.* **1994**, *106*, 1780–1802; *Angew. Chem. Int. Ed. Engl.* **1994**, *33*, 1699–1720; c) V. J. Hruby, G. Li, C. Haskell-Luevano, M. Shenderovich, *Biopolymers* **1997**, *43*, 219–248; d) E. Mann, H. Kessler, *Org. Lett.* **2003**, *5*, 4567–4570.
- [10] a) Y. Kawanaka, E. M. Phillips, K. A. Scheidt, *J. Am. Chem. Soc.* **2009**, *131*, 18028–18029; b) A. Chan, K. A. Scheidt, *J. Am. Chem. Soc.* **2008**, *130*, 2740–2741; c) M. Nahrwold, A. Stončius, A. Penner, B. Neumann, H. G. Stammer, N. Sewald, *Beilstein J. Org. Chem.* **2009**, *5*, 43; d) S. Abele, D. Seebach, *Eur. J. Org. Chem.* **2000**, 1–15, and references therein; e) G. Guichard, S. Able, D. Seebach, *Helv. Chim. Acta* **1998**, *81*, 187–206, and references therein.
- [11] See the Supporting Information for synthetic detail and pertinent literature.
- [12] a) V. V. Lipson, N. Y. Gorobets, *Mol. Diversity* **2009**, *13*, 399–419; b) Y. Oikawa, K. Sugano, O. Yonemitsu, *J. Org. Chem.* **1978**, *43*, 2087–2088.

- [13] a) M. H. El-Badri, M. J. Kurth, *J. Comb. Chem.* **2009**, *11*, 228–238; b) H. Takikawa, K. Hikita, K. Suzuki, *Synlett* **2007**, 2252–2256, and references therein.
- [14] a) G. J. Yu, C. L. Yoo, B. Yang, M. W. Lodewyk, L. Meng, T. T. El-Idreesy, J. C. Fettinger, D. J. Tantillo, A. S. Verkman, M. J. Kurth, *J. Med. Chem.* **2008**, *51*, 6044–6054; b) C. L. Yoo, G. J. Yu, B. Yang, L. I. Robins, A. S. Verkman, M. J. Kurth, *Bioorg. Med. Chem. Lett.* **2008**, *18*, 2610–2614.
- [15] a) F. Dosselaere, J. Vanderleyden, *Crit. Rev. Microbiol.* **2001**, *27*, 75–131; b) F. Roberts, W. C. Roberts, J. J. Johnson, D. E. Kyle, T. Krell, J. R. Coggins, G. H. Cooms, W. K. Milhous, S. Tzipori, D. J. Ferguson, D. Chakrebarati, R. McLeod, *Nature* **1998**, *393*, 801–805.

Received: May 28, 2010
Published online: July 9, 2010